GANPAT UNIVERSITY

B. Tech. Semester: Vth Mechanical Engineering Regular Examination Nov – Dec 2015

2ME501 Material Technology

Time: 3	Hours			O,

1	ime:	3 Hours Total Marks	: 70
ln	struct	ions: i) Answer two sections separately. ii) Figures to the right indicate full marks. iii) Assume suitable data if necessary.	
		Section I	
Q.	1 [a] Explain Gibbs phase rule.	12
	[b	What is cooling curve? Explain cooling curve for binary alloy forming solid solution and binary alloy forming eutectic.	4
	[e]	Explain importance of lever arm principle. Find out percentage of micro constituent presents at 0.2% C, 0.4%C. 0.6% C and 1.4% C at room temperature.	4
0.1		OR	
Q.1	[a]	refractories for furnaces and Cupola	12 4
	[b]	Enlist methods of polymer materials processing. Explain any one method in detail.	4
	[c]	Explain role of matrix and reinforcement materials in composite materials.	34
).2	[a]	With the help of suitable diagram, show full annealing and hardening temperature ranges for plain carbon steel in case of hypo eutectoid and hypereutectoid steel respectively. What are the criteria for selection of these temperature ranges?	12 5
	[b]	What do the microstructure of martemperd and austempered steels consist of? Enlist and explain limitations of both processes.	4
	[c]	Differentiate between annealing and normalizing process.	3
2		OR	
2	[a]	What is hardening process? Discuss factors affecting hardening process.	12 5
	[b]	Explain why hardening by quenching is followed by tempering treatment? Describe the structural changes that take place during tempering.	4
	[c]	Explain temper brittleness and temper colour	

0.3			1			
Q.3	[a]	Explain effects of following elements on cast iron: i) Silicon ii) Sulphur iii) Manganese iv) Phosphorus	3			
	[b]	Differentiate between gray and nodular cast iron with respect to microstructure, properties and applications.	4			
	[c]	Differentiate between white and malleable cast iron with respect to microstructure, properties and applications.	4			
		Section - II				
Q.4			11			
	[a]	Define solid solution. Explain Hume Rothery rules.	4			
	[b]	Draw Iron-Iron Carbide phase diagram and the label the phase fields. Discuss the different reactions that take place in the systems.	5			
			2			
	[c]	Define following phases: i) Ferrite, ii) Austenite				
		OR				
Q.4			11			
ν.,	[a]	Enlist and explain constituents and properties of Brass and Bronze.	4			
	[b]	Enlist properties and applications of any two Aluminum alloy.	4 3			
	[c]	Enlist and explain classification of Composite materials.	12			
Q.5	[a]	Enlist advantages, disadvantages and applications of Powder Metallurgy.				
	[b]	Enlist methods of production of powder in Powder metallurgy. Explain any two methods in detail.	4			
	[e]	Explain compacting, presintering and sintering operation in powder Metallurgy.	4			
		OR				
Q.5			12			
Q.o	[a]	Explain procedural steps to construct T.T.T. diagram of 0.8 % carbon steel	4			
	[b]	Draw a neat sketch of T.T.T. diagram for 0.8 % carbon steel and label the phases therein.	4			
	()	Differentiate between hardness and hardenability of steel. Explain Jominy	4			
	[c]	end quench test.				
Q.6		Write short notes on the following:(Any Three)	12			
	[a]					
	[b]					
	[c] [d]	그림에 있는 것 같아요. 이 하지만 하지 않아 하는 것이 하지만 하지만 하지만 하는 것이 없는 것이 없는 것이 되었다. 그리고 있는 것이 없는 것이 없는 것이 없는 것이 없는 것이 없는 것이 없는				
	[e]	Flame hardening				
		End of Paper				

Page 2 of 2