## B.TECH. SEM- V MECHATRONICS ENGINEERING EXAMINATION NOV/DEC-2011 MC- 502 MATERIAL TECHNOLOGY

| Time: 3 I | Hour]      | [Total Mark  | s: 70] |
|-----------|------------|--|--------|
| nstructio |            | PEGUAR EXAMINATION OF VIEW DECEMBER 2011   |        |
|           | ,          | empt all questions.  ume suitable data if necessary.   | 3:     |
|           | ,          | ures to the right indicate full marks.   |        |
| 1.        | ) 115      | (i) Centritugal atomization (ii) Ball milling  |        |
|           |            | Section – I replience against the design considers I — noise section – I   |        |
|           |            | Dische the functions of pre-sintering and sintering process,   |        |
| Que:1     | , and      | Will at the State of the state  | 12     |
|           | (A)<br>(B) | What is cooling curve? Explain cooling curve for binary solid solution alloy. Explain the importance of tie line and lever arm principle with suitable   |        |
|           |            | examples.  |        |
|           | (C)        | Explain allotropic transformation of pure iron.  OR  |        |
|           |            | Wast is horizon explain process variables involved in heat weatmen   | 12     |
| Que:1     | (4)        | Define the following terms:  | 1 has  |
|           | (A)        | (i) Phase (ii) Solid solution (iii) Degree of freedom  |        |
|           | (B)        | Draw a neat sketch and explain following phase diagram for binary eutectic system:   |        |
|           |            | (i) Two metals completely soluble in the liquid state but completely   |        |
|           |            | insoluble in solid state.  |        |
|           |            | (ii) Two metals completely soluble in the liquid state but partially soluble in solid state.   |        |
|           | (C)        | Explain Gibbs phase rule with suitable examples.   |        |
| 0 2       |            | ent ments and demonstrate of the state of th | 11     |
| Que:2     | (A)        | Draw Fe-Fe <sub>3</sub> C equilibrium diagram with all necessary details.  | ,      |
|           | (B)        | Draw microstructure of 0.2, 0.8 and 1.2% carbon steel. Also find relative  |        |
|           |            | amounts of phase present at room temperature.  |        |
|           | (C)        | Define the following terms:  |        |
|           |            | (i) Ferrite (ii) Cementite (iii) Pearlite  |        |
|           |            | on different type OR sommigates.   |        |
| Que:2     |            |  | 11     |
| Que.2     | (A)        | What is the principle of cyaniding? Differentiate between nitriding and cyaniding.   |        |
|           | (B)        | Explain the principle of carburizing. Differentiate between solid carburizing and gas carburizing.   |        |
|           | (C)        | Explain briefly induction hardening.   |        |
|           | 7          | in sand casting process  | 12     |
| Que: 3    |            | TTT 1' Press a part skatch of TTT disgram for 0.8% carbon  | 12     |
|           | (A)        | What is T.T.T. diagram? Draw a neat sketch of T.T.T. diagram for 0.8% carbon steel.  |        |
|           | (B)        | Explain the following with respect to T.T.T. diagram:  |        |
|           |            | (i) Austempering (ii) Martempering (iii) Critical cooling rate.  |        |
|           | (C)        | Define following cast iron with respect to its microstructure and application:   |        |
|           |            | (i) Gray cast iron and (ii) Malleable cast iron.   |        |

## Section - II Que:4 What are the main stages of powder metallurgy? Also give the advantages and (A) limitation of powder metallurgy? Explain in detail the brequetting or die compaction process. (B) Enumerate the main characteristics of metal powder. (C) OR Que:4 Explain the following powder metallurgy process: (A) (i) Centrifugal atomization (ii) Ball milling Discuss the design consideration require for powder metallurgy articles. (B) Describe the functions of pre-sintering and sintering process. (C) Que:5 Write the different techniques of corrosion control. And also discuss the design (A) against the corrosion. Explain the oxygen absorption mechanism with suitable example. (B) Differentiate the electrochemical corrosion and direct corrosion. (C) 11 Que:5 What is heattreatment? Explain process variables involved in heat treatment (A) Differentiate between annealing and normalizing process. Explain importance (B) of isothermal annealing process. What is tempering? Define temper color and temper brittleness. (C) 12 Que: 6 Write short notes on following: (Any three) Define the following terms: (i) Crystalline material(ii) Amorphous material (iii) Space lattice Diffusion annealing (B) Hardening process (C) Induction hardening (D) Flame hardening (E) END OF PAPER