GANPAT UNIVERSITY

M.Tech. [ME-(AMT)] SEM-I CBCS (NEW) Regular Theory Nov.-Dec.-2015 3ME101 MATERIAL SCIENCE

		ions: i) Answer two sections must be separately. ii) Figures to the right indicate full marks. iii) Assume suitable data if necessary.	60
0	r p	Section - I	
Q.	1 [a]	Define Polymerization. Explain the mechanisms of polymerization. Classify polymerization processes.	03
	[b]	Differentiate between addition and condensation polymerization.	03
	[c]	i) Fillers, ii) Plasticizers, iii) Colorants, iv) Stabilizers	04
Q.1		OR	
	[a]	What is Metal Matrix Composites? Enlist and explain matrix and reinforcement materials for MMCs.	03
	[b]	Explain fiberglass-reinforcement composite.	04
	[c]	Write short note on carbon fiber.	03
Q.2	[a]	Define ceramic and explain classification of ceramic materials. Also enlist various characteristics and applications of ceramic materials.	03
	[b]	Enlist different ceramic forming techniques. Explain Slip casting process in details.	04
	[c]	Explain with neat sketch silicate structure in ceramic materials.	03
0.2		OR	
Q.2	[a]	Classify different types of rubbers used in industry and discuss its structure, properties and application of each one.	04
	[b]	Explain interfaces in composite.	03
	[c]	Discuss the Pultrusion process in detail.	03
2.3	[a] [b] [c] [d]	Write short notes on the following:(Any Two) Ceramic matrix composite Pitting corrosion Prevention of Intergrannular corrosion Carbon nanotube	10

Section II

Q.4	اما	What are point defects? Explain various point defects with sketch.	03
		Explain secondary bonding and enlist examples of secondary bonded	04
	[b]	materials in details. What is dislocation? Differentiate between edge and screw dislocation.	03
Q.5	[a]	Differentiate between low angle grain boundary and high angle grain boundary. Also explain twin boundary.	03
	[b]	Enlist two major differences between deformation by twinning and deformation by slip relative to mechanism, conditions of occurrence and final result.	04
*	[c]	Differentiate between ionic, metallic and covalent bonding.	03
Q.5		OR	
Q.C	[a]	Define strengthening mechanism. Explain solid solution strengthening.	04
	[b]	Explain strain hardening. Briefly explain why lead and tin do not strain harden when deformed at room temperature.	03
	[c]	Explain critical resolved shear stress and dislocation climb.	03
Q.6	[a]	Explain strengthening mechanism of solids by marten site strengthening.	04
	[b]	Explain Griffith theory of brittle fracture.	03
	[c]	Write short notes on: Fracture toughness.	03
		OR	
Q.6			
Q.U	[a]	What is creep? Differentiate between Fatigue and Creep.	04
	[p]	Explain stages in the development of ductile fracture. Also differentiate between ductile and brittle fracture.	04
	[c]	How can the fatigue resistance of materials be improved? End of Paper	02