

**GANPAT UNIVERSITY**  
**M.Tech. Sem I (CAD/CAM) Mechanical Engineering**  
**January -2012 Examination**  
**3ME111 Material Science & Technology**

**Time: 3 Hour**

**Total Marks: 70**

- Instructions:** (i) Attempt all questions.  
(ii) Figures to the right indicate full marks.  
(iii) Two sections must be written in separate answer sheets.

**Section I**

- Q-1** **11**
- (a) Enlist objectives of strengthening mechanism. Discuss strengthening by grain size reduction.
  - (b) Explain solid solution strengthening in detail. Also discuss lattice strain due to solute atom.
  - (c) Explain the process of annealing of cold work metal.
- OR**
- Q-1** **11**
- (a) Explain the role of dislocation in plastic deformation of material. Explain the process of strain hardening in light of dislocation theory.
  - (b) Differentiate between edge and screw dislocation.
  - (c) Differentiate between primary and secondary bonding.
- Q-2** **12**
- (a) What is creep? Explain the method of creep test and creep curve in detail.
  - (b) What is fatigue? Explain the mechanism of fracture taking place by fatigue failure.
  - (c) How can the fatigue resistance of materials be improved?
- OR**
- Q-2** **12**
- (a) Explain following with respect to brittle fracture:
    - i) Stress concentration,
    - ii) Fracture toughness
  - (b) Explain the S-N curve and endurance limit for article made of steel.
  - (c) Explain the stages in development of ductile fracture. Also differentiate between ductile and brittle fracture.
- Q-3** **12**
- (a) What is corrosion? Differentiate between dry and wet corrosion. Also discuss factors affecting rate of corrosion.
  - (b) What is pitting corrosion? Differentiate between pitting and crevice corrosion. Explain preventive measure from pitting corrosion.
  - (c) What is intergranular corrosion? Differentiate between weld decay and knife line attack. Explain preventive measure for intergranular corrosion.



## Section II

Q-4

- (a) Explain with neat sketch the atomic bonding arrangement in ~~SiO<sub>2</sub>~~ <sup>SiO<sub>2</sub></sup> tetrahedron, also discuss with suitable examples the island, ring, chain, sheet and network structure of silica. 12
- (b) Discuss the effect of crystalline structure and molecular weight on properties of polymer.
- (c) Why are composites used in place of metals, ceramics and polymers?

OR

Q-4

- (a) Classify polymer based on applications, physical properties, physicochemical reactions and mode of preparation. 12
- (b) Discuss the effect of following on properties of polymer:  
(i) Fillers, (ii) Plasticizers, (iii) Colorants and (iv) Stabilizers.
- (c) Discuss the method of polymerization.

Q-5

- (a) Explain the characteristics of polymers. 11
- (b) Discuss the reaction mechanism of chain addition polymerization.
- (c) Explain Branched and cross linked polymers.

OR

Q-5

- (a) Explain following polymer structure with suitable examples: 11  
(i) Linear polymers,  
(ii) Network polymer
- (b) Define the term 'Traditional ceramics' and 'modern ceramics'. How traditional ceramics differ from modern ceramics.
- (c) Write short note on: Mechanical properties of ceramic materials.

Q-6

- Write short note on any three : 12
- (i) Burger Vector
- (ii) Differentiate between thermoplastic and thermosetting
- (iii) Defects arising in solid materials.
- (iv) Polymer Matrix Composite

End of Paper